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- 4 24. (Previously presented) The process of claim 22, further comprising the steps of:
- 5 heating the reaction mixture to reflux; and
- 6 cooling the reaction mixture to room temperature.

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- 8 25. (Previously presented) The process of claim 24, further comprising isolating the POSS
- 9 fragment.
- 1 26. (Previously presented) The process of claim 25, wherein the POSS fragment is isolated
- 2 by distillation, filtration, evaporation, decantation, crystallization, pressure reduction, or
- 3 extraction, or a combination thereof.

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- 5 27. (Previously presented) The process of claim 26, further comprising the step of purifying
- 6 the isolated POSS fragment through washing with water.
- 1 28. (Previously presented) The process of claim 22 wherein the base cleaves at least one
- 2 silicon-oxygen-silicon (Si-O-Si) bond in the polymeric silsesquioxane to promote the conversion
- 3 of the polymeric silsesquioxane into the POSS fragment.
- 1 29. (Amended) The process of claim 28, wherein the base is selected from the group
- 2 consisting of hydroxide [OH], organic alkoxides [R"O], carboxylates [R"COO], amides
- 3 $[R''NH]^{-}$, carboxamides $[R''C(O)NR'']^{-}$, carboxamides $[R'']^{-}[[,]]$, carboxamides $[CO_3]^{-2}$, sulfate $[SO_4]^{-2}$,
- 4 phosphate [PO₄]⁻³, biphosphate [HPO₄]⁻², phosphorus ylides [R"₄P], nitrate [NO₃], borate
- 5 [B(OH)₄], cyanate [OCN], fluoride [F], hypochlorite [OCl], silicate [SiO₄], stannate [SnO₄],
- 6 basic metal oxides comprising Al₂O₃, CaO, and ZnO, amines R"₃N and amine oxides R"₃NO,
- and organometallics comprising R"Li, R"₂Zn, R"₂Mg, and R"MgX', where R" represents an
- 8 organic substituent and multiple organic substituents need not be identical, and X' represents an

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9 inorganic substituent.

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- 1 30. (Previously presented) The process of claim 22, wherein a mixture of different bases is
- 2 used.
- 1 31. (Previously presented) The process of claim 22, further comprising mixing a co-reagent
- with the base and the polymeric silsesquioxane in the solvent.
- 1 · 32. (Previously presented) The process of claim 31, wherein the co-reagent is selected from
- 2 the group consisting of common Grignard reagents [[R"MgX]] R"MgX', alkalihalides, zinc
- 3 compounds comprising ZnI₂, ZnBr₂, ZnCl₂, and ZnF₂, aluminum compounds comprising Al₂H₆,
- 4 LiAlH₄, AlI₃, AlBr₃, AlCl₃, and AlF₃, and boron compounds comprising R"B(OH)₂, BI₃, BBr₃,
- 5 BCl₃, and BF₃, where R" represents an organic substituent and X' represents an inorganic
- 6 substituent.

Claims 33-45 (cancelled).

- 1 46. (Previously presented) A process of converting a plurality of POSS fragments into a
- 2 POSS compound, comprising:
- mixing an effective amount of a base with the plurality of POSS fragments in a solvent to
- 4 produce a basic reaction mixture, the base reacting with the POSS fragments to produce the
- 5 POSS compound,
- wherein the POSS fragments have the formula (RSiO_{1.5})_m(RXSiO_{1.0})_n and contain from 1
- 7 to 7 silicon atoms and no more than 3 rings, and the POSS compound is selected from the group
- 8 consisting of homoleptic nanostructure compounds having the formula $[(RSiO_{1.5})_n]_{\Sigma\#}$
- 9 heteroleptic nanostructure compounds having the formula $[(RSiO_{1.5})_m(R'SiO_{1.5})_n]_{\Sigma\#}$
- 10 functionalized homoleptic nanostructure compounds having the formula
- 11 $[(RSiO_{1.5})_m(RXSiO_{1.0})_n]_{\Sigma^{\#}}$, functionalized heteroleptic nanostructure compounds having the
- formula $[(RSiO_{1.5})_m(R'SiO_{1.5})_n(RXSiO_{1.0})_p]_{\Sigma^\#}$, and expanded POSS fragments having the
- 13 formula (RSiO_{1.5})_m(RXSiO_{1.0})_n, where R and R' each represents an organic substituent, X
- represents a functionality substituent, m, n and p represent the stoichiometry of the formula, \sum

- indicates nanostructure, and # represents the number of silicon atoms contained within the
- 16 nanostructure.
- 1 47. (Previously presented) The process of claim 46, wherein the base and the POSS
- 2 fragments are mixed by stirring the reaction mixture.
- 1 · 48. (Previously presented) The process of claim 46, further comprising the steps of:
- 2 heating the reaction mixture to reflux; and
- 3 cooling the reaction mixture to room temperature.
- 1 49. (Previously presented) The process of claim 48, further comprising:
- 2 isolating the POSS compound.
- 1 50. (Previously presented) The process of claim 49 wherein the POSS compound is isolated
- 2 by distillation, filtration, evaporation, decantation, crystallization, pressure reduction, or
- 3 extraction, or a combination thereof.
- 1 51. (Previously presented) The process of claim 50, further comprising the step of purifying
- 2 the isolated POSS compound through washing with water.
- 1 52. (Previously presented) The process of claim 46, wherein the base cleaves at least one
- 2 silicon-oxygen-silicon (Si-O-Si) bond in the POSS fragments to promote the conversion of the
- 3 POSS fragments into the POSS compound.
- 1 53. (Previously presented) The process of claim 52, wherein the base is selected from the
- 2 group consisting of hydroxide [OH], organic alkoxides [R"O], carboxylates [R"COO], amides
- 3 [R"NH], carboxamides [R"C(O)NR"], carbanions [R"], carbonate [CO₃], sulfate [SO₄],
- 4 phosphate $[PO_4]^{-3}$, biphosphate $[HPO_4]^{-2}$, phosphorus ylides $[R''_4P]^{-1}$, nitrate $[NO_3]^{-1}$, borate
- 5 [B(OH)₄], cyanate [OCN], fluoride [F], hypochlorite [OCl], silicate [SiO₄]⁻⁴, stannate [SnO₄]⁻⁴,
- 6 basic metal oxides comprising Al₂O₃, CaO, and ZnO, amines R"₃N and amine oxides R"₃NO,

- 7 and organometallics comprising R"Li, R"₂Zn, R"₂Mg, and R"MgX', where R" represents an
- 8 organic substituent and multiple organic substituents need not be identical, and X' represents an
- 9 inorganic substituent.
- 1 54. (Previously presented) The process of claim 53, wherein the concentration of the base is
- 2 between 1-10 equivalents per mole of silicon present in the reaction mixture.
- 1 55. (Previously presented) The process of claim 54, wherein the concentration of the
- 2 hydroxide base is between 1-2 equivalents per mole of silicon present in the reaction mixture.
- 1 56. (Previously presented) The process of claim 46, wherein a mixture of different bases is
- 2 used.
- 1 57. (Previously presented) The process of claim 46, further comprising mixing a co-reagent
- with the base and the plurality of POSS fragments in the solvent.
- 1 58. (Amended) The process of claim 47, wherein the co-reagent is selected from the group
- 2 consisting of common Grignard reagents [[R"MgX]] R"MgX', alkalihalides, zinc compounds
- 3 comprising ZnI₂, ZnBr₂, ZnCl₂, and ZnF₂, aluminum compounds comprising Al₂H₆, LiAlH₄,
- 4 AlI₃, AlBr₃, AlCl₃, and AlF₃, and boron compounds comprising R"B(OH)₂, BI₃, BBr₃, BCl₃,
- 5 and BF₃, where R" represents an organic substituent and X' represents an inorganic substituent.
- 1 59. (Previously presented) A process of converting a first functionalized POSS
- 2 nanostructure compound into a second functionalized POSS nanostructure compound that is
- 3 different than the first functionalized POSS nanostructure compound, comprising:
- 4 mixing an effective amount of a base with the first functionalized POSS nanostructure
- 5 compound in a solvent to produce a basic reaction mixture, the base reacting with the first
- 6 functionalized POSS nanostructure compound to produce the second POSS nanostructure

7 compound,

- wherein the first and second POSS nanostructure compounds are each selected from the group consisting of homoleptic nanostructure compounds having the formula $[(RSiO_{1.5})_n]_{\Sigma}$ #,
- 10 heteroleptic nanostructure compounds having the formula $[(RSiO_{1.5})_m(R'SiO_{1.5})_n]_{\Sigma}$,
- 11 functionalized homoleptic nanostructure compounds having the formula
- 12 $[(RSiO_{1.5})_m(RXSiO_{1.0})_n]_{\Sigma^{\#}}$, and functionalized heteroleptic nanostructure compounds having the
- formula $[(RSiO_{1.5})_m(R'SiO_{1.5})_n(RXSiO_{1.0})_p]_{\Sigma^\#}$, where R and R' each represents an organic
- 14 substituent, X represents a functionality substituent, m, n and p represent the stoichiometry of the
- 15 formula, \sum indicates nanostructure, and # represents the number of silicon atoms contained
- 16 within the nanostructure.
- 1 60. (Previously presented) The process of claim 59, wherein the second functionalized POSS
- 2 nanostructure compound has more functionalities X than the first functionalized POSS
- 3 nanostructure compound but the two functionalized POSS nanostructure compounds have the
- 4 same number of silicon atoms.
- 1 61. (Previously presented) The process of claim 59, wherein the base and the first
- 2 functionalized POSS nanostructure compound are mixed by stirring the reaction mixture.
- 1 62. (Previously presented) The process of claim 61, further comprising the steps of:
- 2 heating the reaction mixture to reflux; and
- 3 cooling the reaction mixture to room temperature.
- 1 63. (Previously presented) The process of claim 62, further comprising:
- 2 isolating the second functionalized POSS nanostructure compound.
- 1 64. (Previously presented) The process of claim 63, wherein the second functionalized POSS
- 2 nanostructure compound is isolated by distillation, filtration, evaporation, decantation,
- 3 crystallization, pressure reduction, or extraction, or a combination thereof.

- 1 65. (Previously presented) The process of claim 64, further comprising the step of purifying
- 2 the isolated POSS nanostructure compound through washing with water.
- 1 66. (Previously presented) The process of claim 59, wherein the base cleaves at least one
- 2 silicon-oxygen-silicon (Si-O-Si) bond in the first functionalized POSS nanostructure compound
- 3 to promote the conversion of the first functionalized POSS nanostructure compound into the
- 4 second functionalized POSS nanostructure compound.
- 1 67. (Previously presented) The process of claim 66, wherein the base is selected from the
- 2 group consisting of hydroxide [OH], organic alkoxides [R"O], carboxylates [R"COO], amides
- 3 [R"NH], carboxamides [R"C(O)NR"], carbanions [R"], carbonate [CO₃], sulfate [SO₄],
- 4 phosphate [PO₄]⁻³, biphosphate [HPO₄]⁻², phosphorus ylides [R"₄P]⁻, nitrate [NO₃]⁻, borate
- 5 [B(OH)₄], cyanate [OCN], fluoride [F], hypochlorite [OCl], silicate [SiO₄]⁻⁴, stannate [SnO₄]⁻⁴,
- basic metal oxides comprising Al₂O₃, CaO, and ZnO, amines R"₃N and amine oxides R"₃NO.
- 7 and organometallics comprising R"Li, R"₂Zn, R"₂Mg, and R"MgX', where R" represents an
- 8 organic substituent and multiple organic substituents need not be identical, and X' represents an
- 9 inorganic substituent.
- 1 70. (Previously presented) The process of claim 59, wherein a mixture of different bases is
- 2 used.
- 1 71. (Previously presented) The process of claim 59, further comprising mixing a co-reagent
- 2 with the base and the first functionalized POSS nanostructure compound in the solvent.
- 1 72. (Amended) The process of claim 71, wherein the co-reagent is selected from the group
- 2 consisting of common Grignard reagents [[R"MgX]] R"MgX', alkalihalides, zinc compounds
- 3 comprising ZnI₂, ZnBr₂, ZnCl₂, and ZnF₂, aluminum compounds comprising Al₂H₆, LiAlH₄,
- 4 AlI₃, AlBr₃, AlCl₃, and AlF₃, and boron compounds comprising R"B(OH)₂, BI₃, BBr₃, BCl₃,
- 5 and BF₃, where R" represents an organic substituent and X' represents an inorganic substituent.

Claims 73-85 (cancelled).

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- 86. (Previously presented) A process of converting an unfunctionalized POSS nanostructure
 compound into a functionalized POSS nanostructure compound, comprising:
 - mixing an effective amount of a base with the unfunctionalized POSS nanostructure compound in a solvent to produce a basic reaction mixture, the base reacting with the unfunctionalized POSS nanostructure compound to produce the functionalized POSS nanostructure compound,
 - wherein the unfunctionalized POSS nanostructure compound is selected from the group consisting of homoleptic nanostructure compounds having the formula $[(RSiO_{1.5})_n]_{\Sigma^{\#}}$ and heteroleptic nanostructure compounds having the formula $[(RSiO_{1.5})_m(R'SiO_{1.5})_n]_{\Sigma^{\#}}$, and the functionalized POSS nanostructure compound is selected from the group consisting of functionalized homoleptic nanostructure compounds having the formula $[(RSiO_{1.5})_m(RXSiO_{1.0})_n]_{\Sigma^{\#}}$ and functionalized heteroleptic nanostructure compounds having the formula $[(RSiO_{1.5})_m(RXSiO_{1.5})_n(RXSiO_{1.0})_p]_{\Sigma^{\#}}$, where R and R' each represents an organic substituent, X represents a functionality substituent, m, n and p represent the stoichiometry of the formula, Σ indicates nanostructure, and π represents the number of silicon atoms contained within the nanostructure.
- 1 87. (Previously presented) The process of claim 86, wherein the base and the unfunctionalized POSS nanostructure compound are mixed by stirring the reaction mixture.
- 1 88. (Previously presented) The process of claim 86, further comprising the steps of:
- 2 heating the reaction mixture to reflux; and
- 3 cooling the reaction mixture to room temperature.
- 1 89. (Previously presented) The process of claim 88, further comprising:
- 2 isolating the functionalized POSS nanostructure compound.

- 1 90. (Previously presented) The process of claim 89, wherein the functionalized POSS
- 2 nanostructure compound is isolated by distillation, filtration, evaporation, decantation,
- 3 crystallization, pressure reduction, or extraction, or a combination thereof.
- 1 91. (Previously presented) The process of claim 90, further comprising the step of purifying
- 2 the isolated functionalized POSS nanostructure compound through washing with water.
- 1 92. (Previously presented) The process of claim 86, wherein the base cleaves at least one
- 2 silicon-oxygen-silicon (Si-O-Si) bond in the unfunctionalized POSS nanostructure compound to
- 3 promote the conversion of the polymeric silsesquioxane into the functionalized POSS
- 4 nanostructure compound.
- 1 93. (Amended) The process of claim 92, wherein the base is selected from the group
- 2 consisting of hydroxide [OH], organic alkoxides [R"O], carboxylates [R"COO], amides
- [R"NH], carboxamides [R"C(O)NR"], carbanions [R"] [[,]], carbonate $[CO_3]^2$, sulfate $[SO_4]^2$,
- 4 phosphate [PO₄]⁻³, biphosphate [HPO₄]⁻², phosphorus ylides [R"₄P]⁻, nitrate [NO₃]⁻, borate
- 5 [B(OH)₄], cyanate [OCN], fluoride [F], hypochlorite [OCl], silicate [SiO₄]⁻⁴, stannate [SnO₄]⁻⁴,
- 6 basic metal oxides comprising Al₂O₃, CaO, and ZnO, amines R"₃N and amine oxides R"₃NO,
- 7 and organometallics comprising R"Li, R"₂Zn, R"₂Mg, and R"MgX', where R" represents an
- 8 organic substituent and multiple organic substituents need not be identical, and X' represents an
- 9 inorganic substituent.
- 1 94. (Previously presented) The process of claim 93, wherein the base is a hydroxide and the
- 2 concentration of the hydroxide base is between 1-10 equivalents per mole of silicon present in
- 3 the reaction mixture.
- 1 95. (Previously presented) The process of claim 94, wherein the concentration of the
- 2 hydroxide base is between 2-5 equivalents per mole of silicon present in the reaction mixture.

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- 1 96. (Previously presented) The process of claim 95, wherein a mixture of different bases is
- 2 used.
- 1 97. (Previously presented) The process of claim 86, further comprising mixing a co-reagent
- with the base and the unfunctionalized POSS nanostructure compound in the solvent.
- 1. 98. (Amended) The process of claim 97, wherein the co-reagent is selected from the group
- 2 consisting of common Grignard reagents [[R"MgX]] R"MgX', alkalihalides, zinc compounds
- 3 comprising ZnI₂, ZnBr₂, ZnCl₂, and ZnF₂, aluminum compounds comprising Al₂H₆, LiAlH₄,
- 4 AlI₃, AlBr₃, AlCl₃, and AlF₃, and boron compounds comprising R"B(OH)₂, BI₃, BBr₃, BCl₃,
- 5 and BF₃, where R" represents an organic substituent and X' represents an inorganic substituent.

Claims 99-113 (cancelled).

- 1 114. (Amended) A process of converting a polymeric silsesquioxane into a POSS
- 2 nanostructure compound, comprising:
- mixing an effective amount of a base with the polymeric silsesquioxane in a solvent to
- 4 produce a basic reaction mixture, the base reacting with the polymeric silsesquioxane to produce
- 5 the POSS nanostructure compound,
- wherein the polymeric silsesquioxane has the formula $[RSiO_{1.5}]_{\infty}$, and the POSS
- 7 nanostructure compound is $[(RSiO_{1.5})_4(RXSiO_{1.0})_3]_{\Sigma 7}$, where R represents an organic substituent,
- 8 X represents a functionality substituent, ∞ represents the degree of polymerization and is a
- 9 number greater than or equal to 1, m, n and p represent the stoichiometry of the formula, and \sum
- 10 indicates nanostructure, , and # represents the number of silicon atoms contained within the
- 11 nanostructure.
- 1 115. (Previously presented) The process of claim 46, wherein the POSS compound is
- 2 $[RSiO_{1.5})_4(RXSiO_{1.0})_3]_{\Sigma 7}$.

- 1 116. (Previously presented) The process of claim 59, wherein the second functionalized POSS
- 2 nanostructure compound is $[(RSiO_{1.5})_4(RXSiO_{1.0})_3]_{\Sigma7}$.

Claim 117 (cancelled).

- 1 118 (Previously presented) The process of claim 86, wherein the functionalized POSS
- 2 ananostructure compound is $[(RSiO_{1.5})_4(RXSiO_{1.0})_3]_{\Sigma^7}$.

Claims 119-134 (cancelled).

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